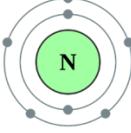


	N_2
Atomic Number	7
Ground State Electron Configuration	$1s^2 2s^2 2p^3$
Number of Valence Electrons	5
Lewis Dot Structure	$\cdot\ddot{\text{N}}\cdot$
Bohr Diagram	
Draw the Ground State Electron Configuration	$ \begin{array}{ccccc} 2p^3 & 1 & 1 & 1 \\ 2s^2 & \underline{1} & & \\ 1s^2 & \underline{1} & & \\ & N & & \\ & \text{Ground State} & & \end{array} $
Hybridization in N_2	sp
Draw the Hybrid State Electron Configuration in N_2	$ \begin{array}{ccccc} 2p^2 & 1 & 1 \\ 2sp^2 & \underline{1} & 1 \\ 1s^2 & \underline{1} & \\ & N & \\ & \text{Hybrid State} & \end{array} $

Draw the orbital diagram for
 N_2

